



CERN, TA1/SSD **coffee-meeting** 11.6.2004

Defects in silicon crystals

- Some very basic remarks -



Michael Moll

CERN - Geneva - Switzerland

People are like crystals. It is the defects in them that make them interesting. Sir F.Charles Frank

Perfection has one grave defect : it is apt to be dull
William Somerset Maugham

- Defects give silicon crystals new properties
- These properties can be useful (e.g. doping, defect engineering) or not (e.g. radiation damage, metal contaminations)



Outline

- **Silicon and Silicon crystal structure**
- **Defect types in silicon crystals**
- **Silicon doping (Dopants = Defects)**
- **Radiation induced defects**
- **What are defects doing to detectors?**
- **How to measure defects?**
- **Coffee**

**Easy
level**

**More
difficult
level**

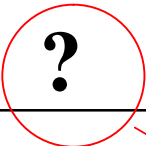
Relax level



Atomic number = 14

Atomic mass 28.0855 amu

- Most abundant solid element on earth 50% O, 26% Si, 8% Al, 5% Fe, 3% Ca, ...
- 90% of earth's crust is composed of silica (SiO_2) and silicate !



Very pure sand or quartz



Silicon

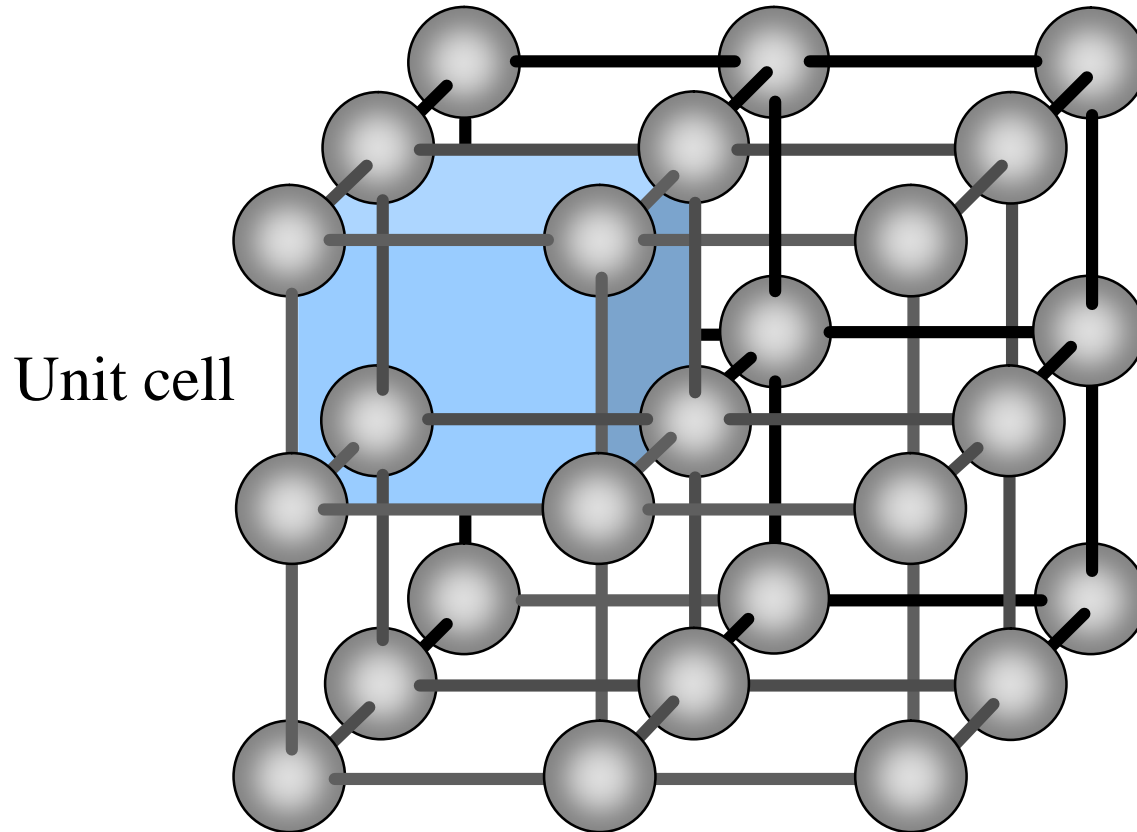
(e.g. from Australia or Nova Scotia in Canada)

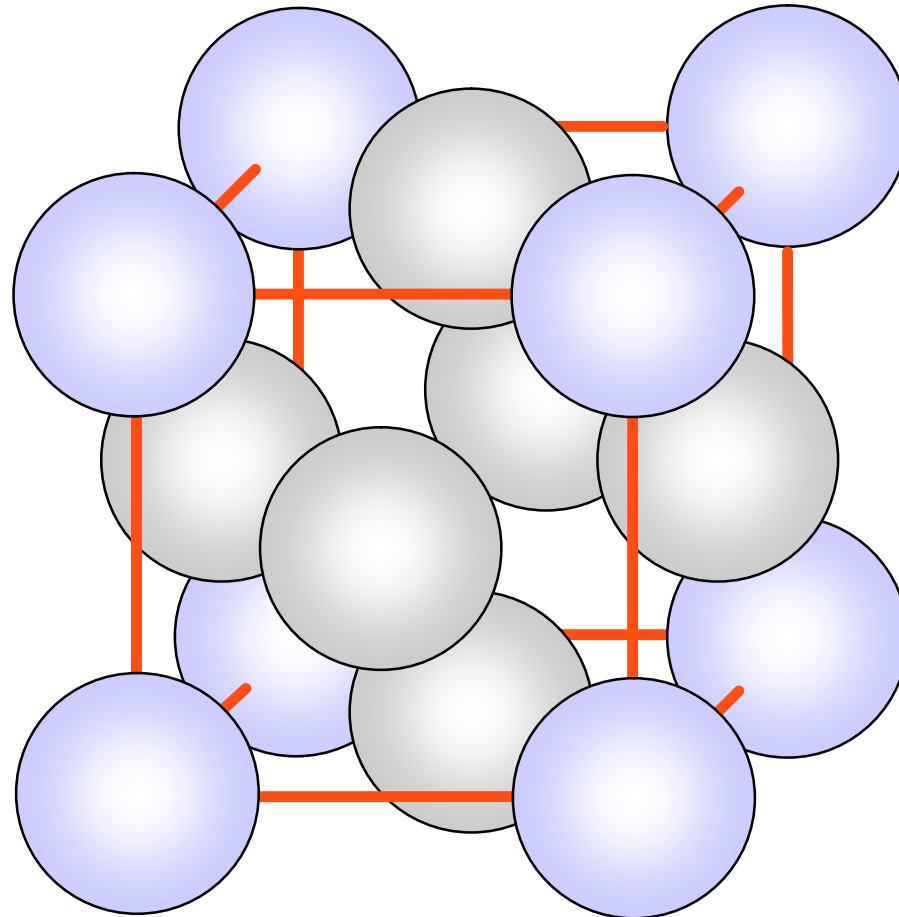
≈ 99% SiO_2

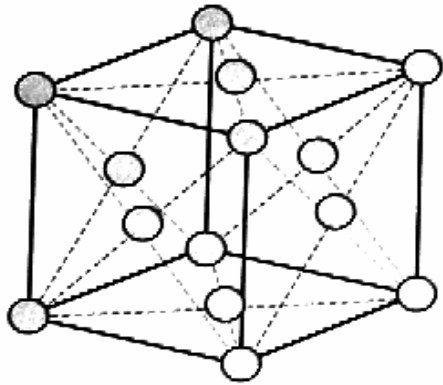
≈ 1% Al_2O_3 , Fe_2O_3 , TiO_2 (CaO , MgO)

Another coffee meeting ?

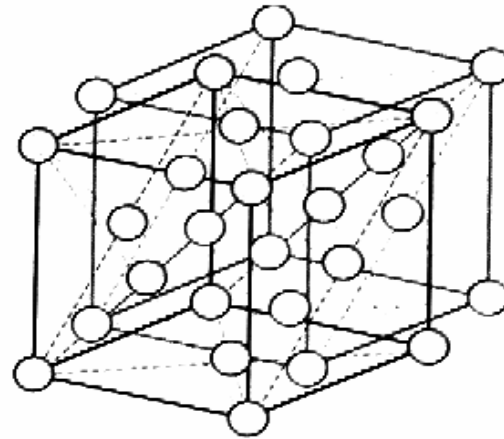
Unit Cell in 3-D Structure



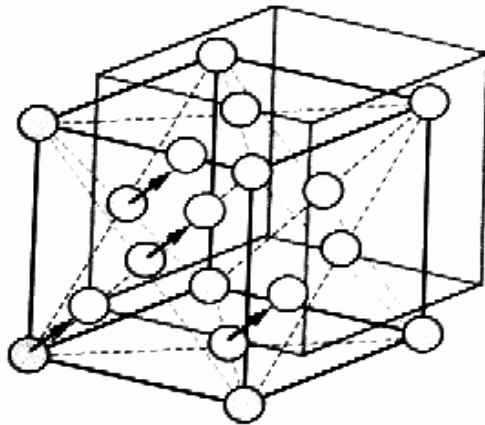




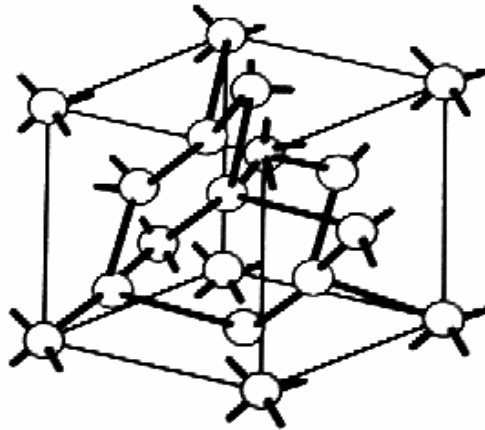
Basic FCC Cell



Merged FCC Cells

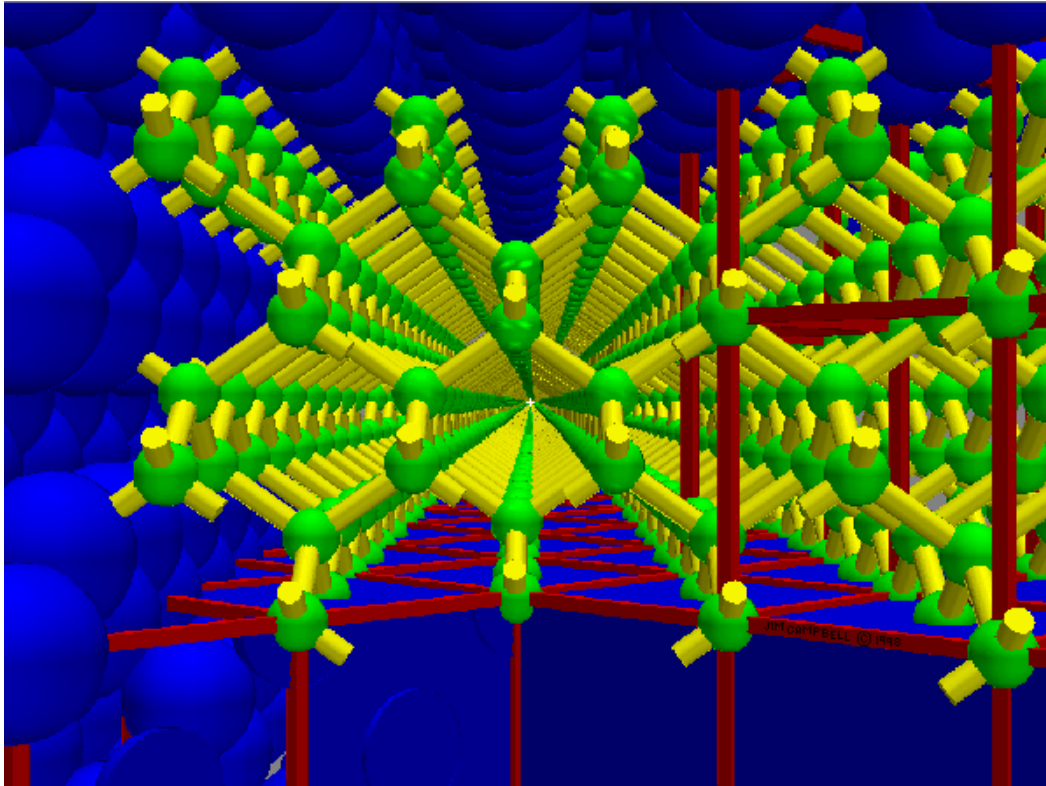


Omitting atoms
outside Cell



Bonding of Atoms

**Silicon has the basic
diamond crystal
structure:
two merged FCC cells
offset by $a/4$ in x, y
and z**

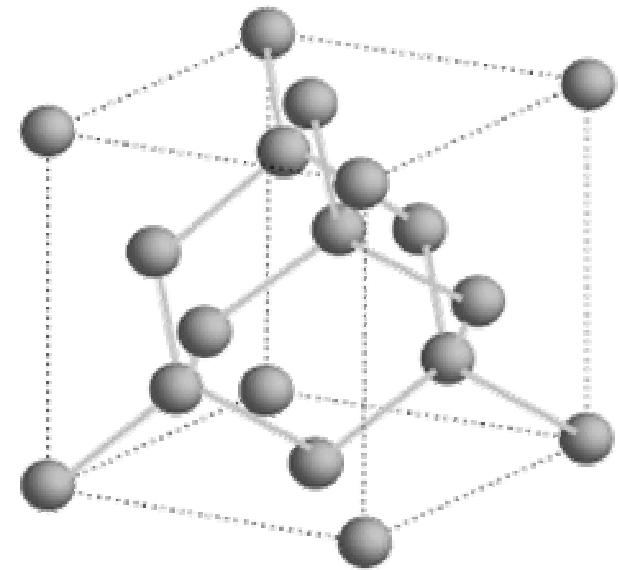


Silicon crystallizes in the same pattern as **Diamond**, in a structure called "two interpenetrating face-centered cubic" primitive lattices. The lines between silicon atoms in the lattice illustration indicate nearest-neighbor bonds. The cube side for silicon is 0.357 nm.

[Link: BC8 Structure of Silicon](#)

[Link: Silicon lattice \(wrl\)](#)

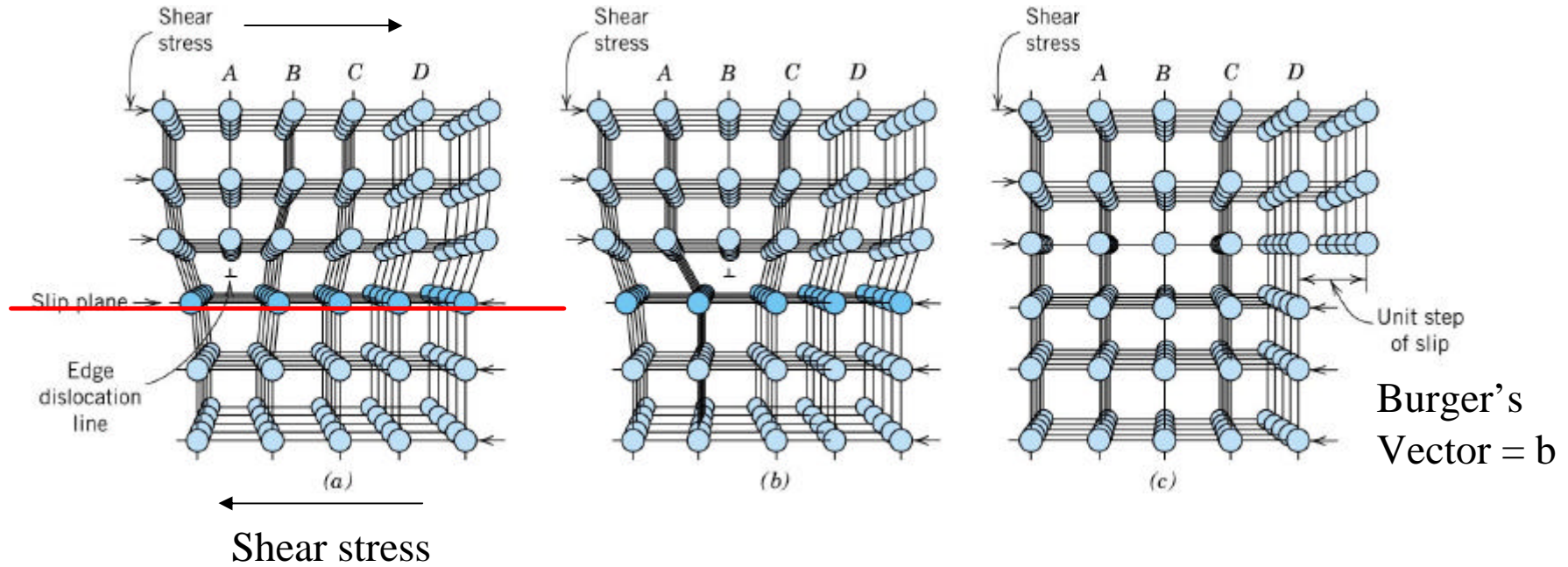
[Link: Silicon lattice with bonds \(wrl\)](#)



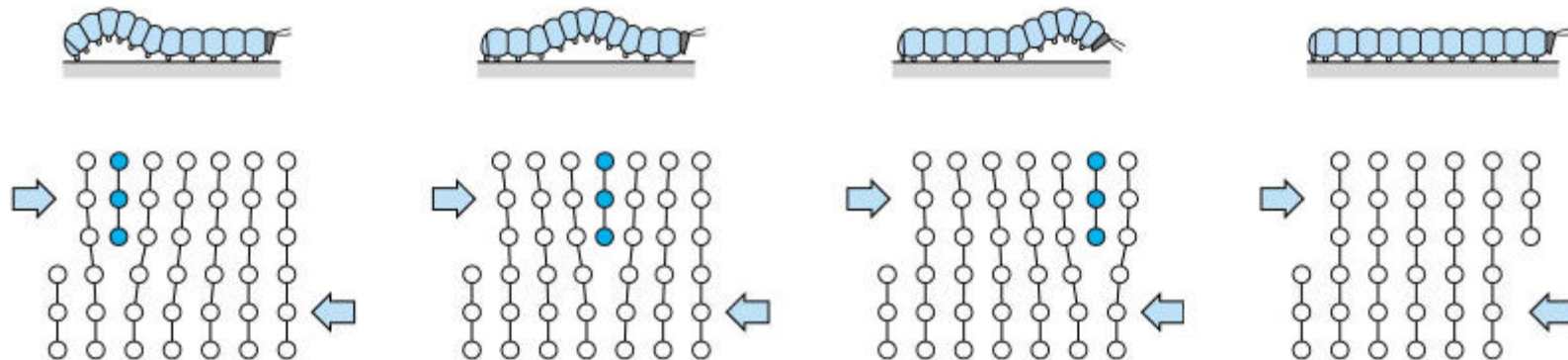
Outline

- Silicon and Silicon crystal structure
- **Defect types in silicon crystals**
 - Lattice defects (Dislocations)
 - Point defects (e.g. Impurities)
 - Cluster defects and Precipitates
- Silicon doping (Dopants = Defects)
- Radiation induced defects
- What are defects doing to detectors?
- How to measure defects?
- Coffee

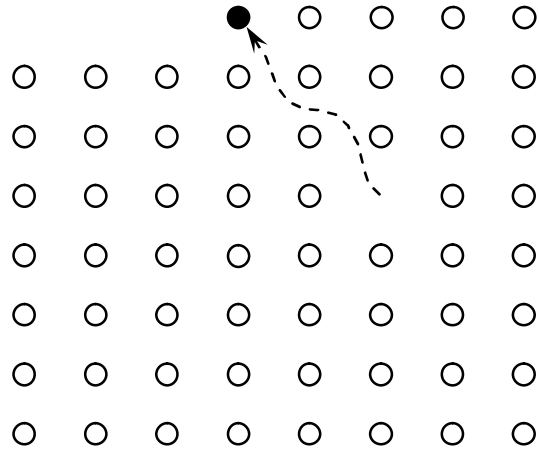




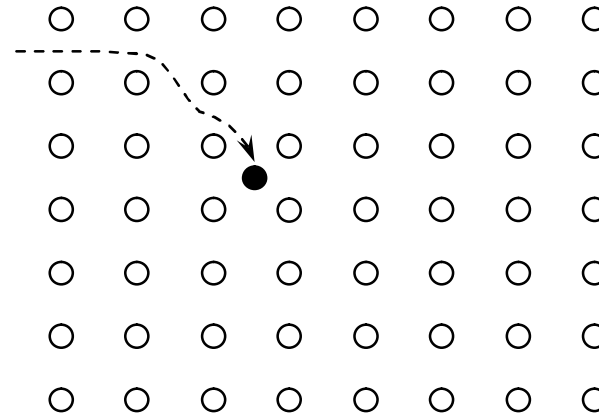
The caterpillar or rug-moving analogy



Point Defects

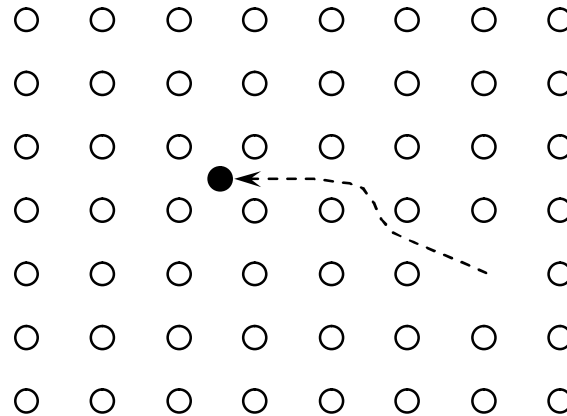


Vacancy defect



Interstitial defect

[Link: Split-interstitial \(wrl\)](#)

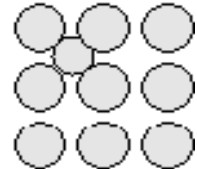
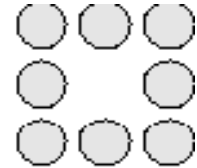


Frenkel defect



- **Intrinsic defects:**

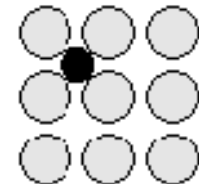
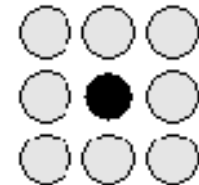
- The **Vacancy (denoted V)**: an atom is removed.
- The **Self-interstitial (denoted I)**: a host atom sits in a normally unoccupied site or interstice (various sites: bond centres, tetrahedral sites, interstitial + displaced regular atom).

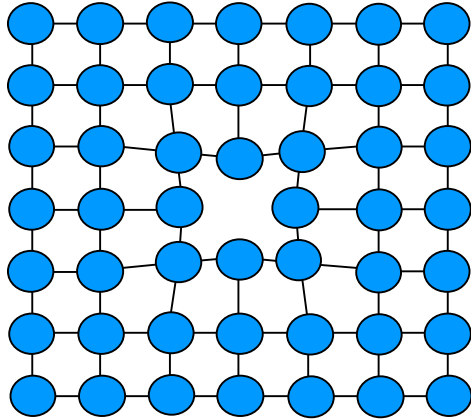


- **Extrinsic defects: due to an impurity.**

These can be:

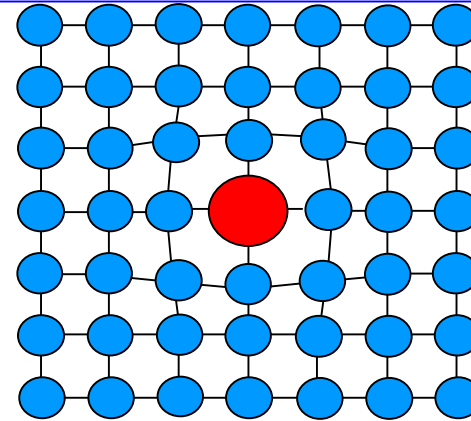
- **Substitutional**, such as carbon substitutional (denoted C_s)
- **Interstitial** (such as the carbon interstitial (denoted C_i)).





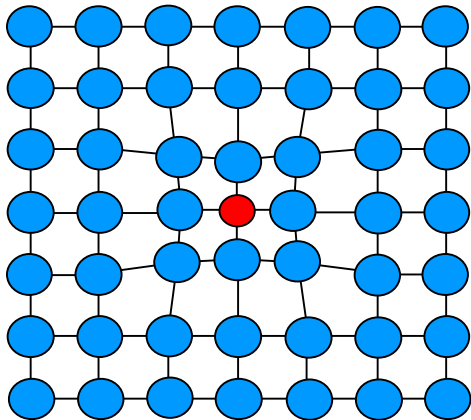
(a) A vacancy in the crystal.

Link: [Vacancy - Hydrogen Defect](#)



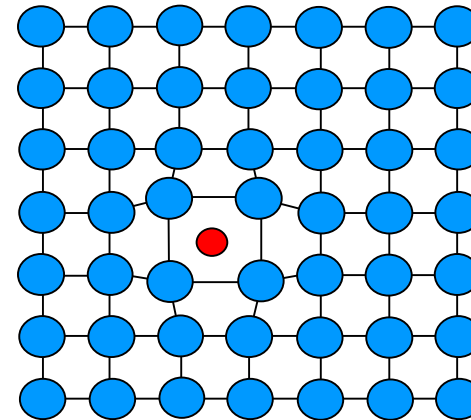
Ge, Sn

(b) A substitutional impurity in the crystal. The impurity atom is larger than the host atom.



C_s

(c) A substitutional impurity in the crystal. The impurity atom is smaller than the host atom.



C_i, O_i

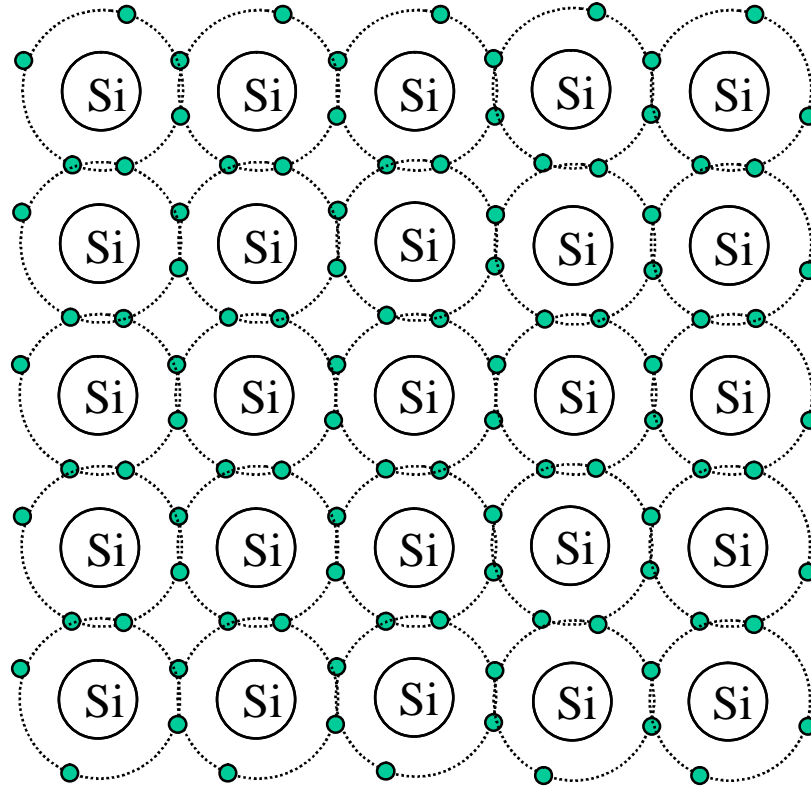
(d) An interstitial impurity in the crystal. It occupies an empty space between host atoms.

Outline

- Silicon and Silicon crystal structure
- Defect types in silicon crystals
- **Silicon doping (Dopants = Defects)**
 - **Intrinsic silicon**
 - **n-type silicon**
 - **p-type silicon**
- Radiation induced defects
- What are defects doing to detectors?
- How to measure defects?
- Coffee

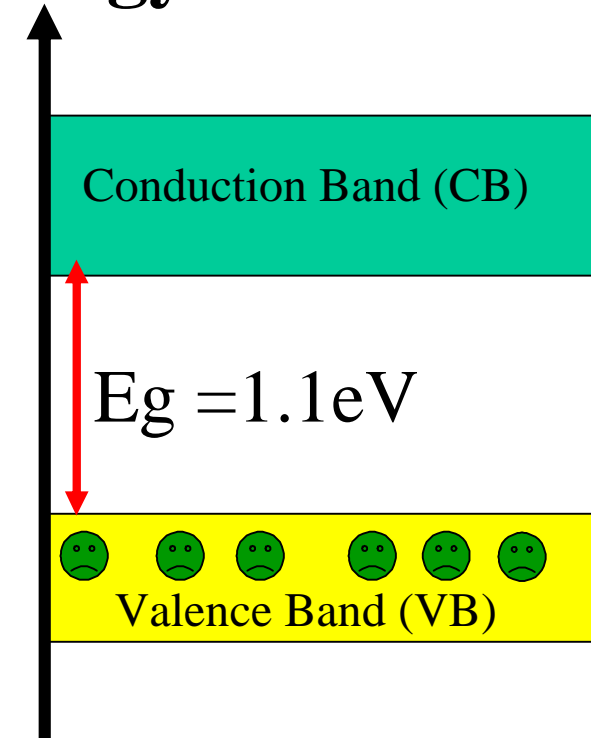


Covalent Bonding of Pure Silicon



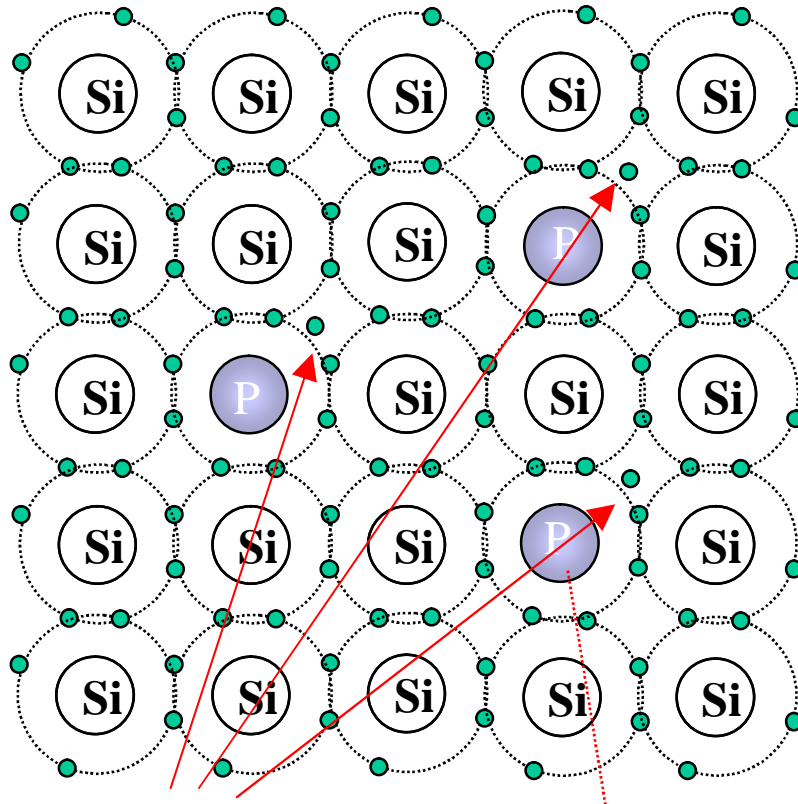
Silicon atoms share valence electrons to form insulator-like bonds.

Energy



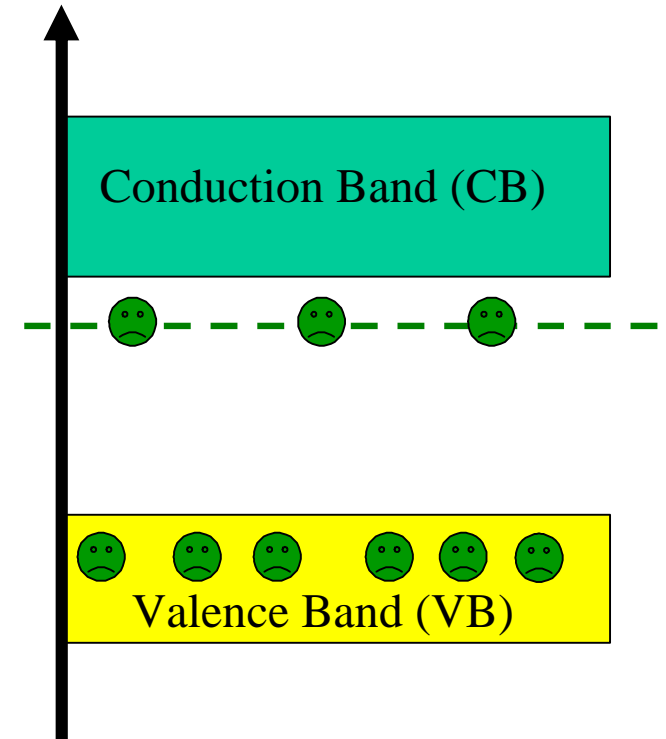


Donor atoms provide excess electrons
to form n-type silicon.



Excess electron (-)

**Phosphorus atom
serves as n-type
dopant**



Conduction in n-Type Silicon

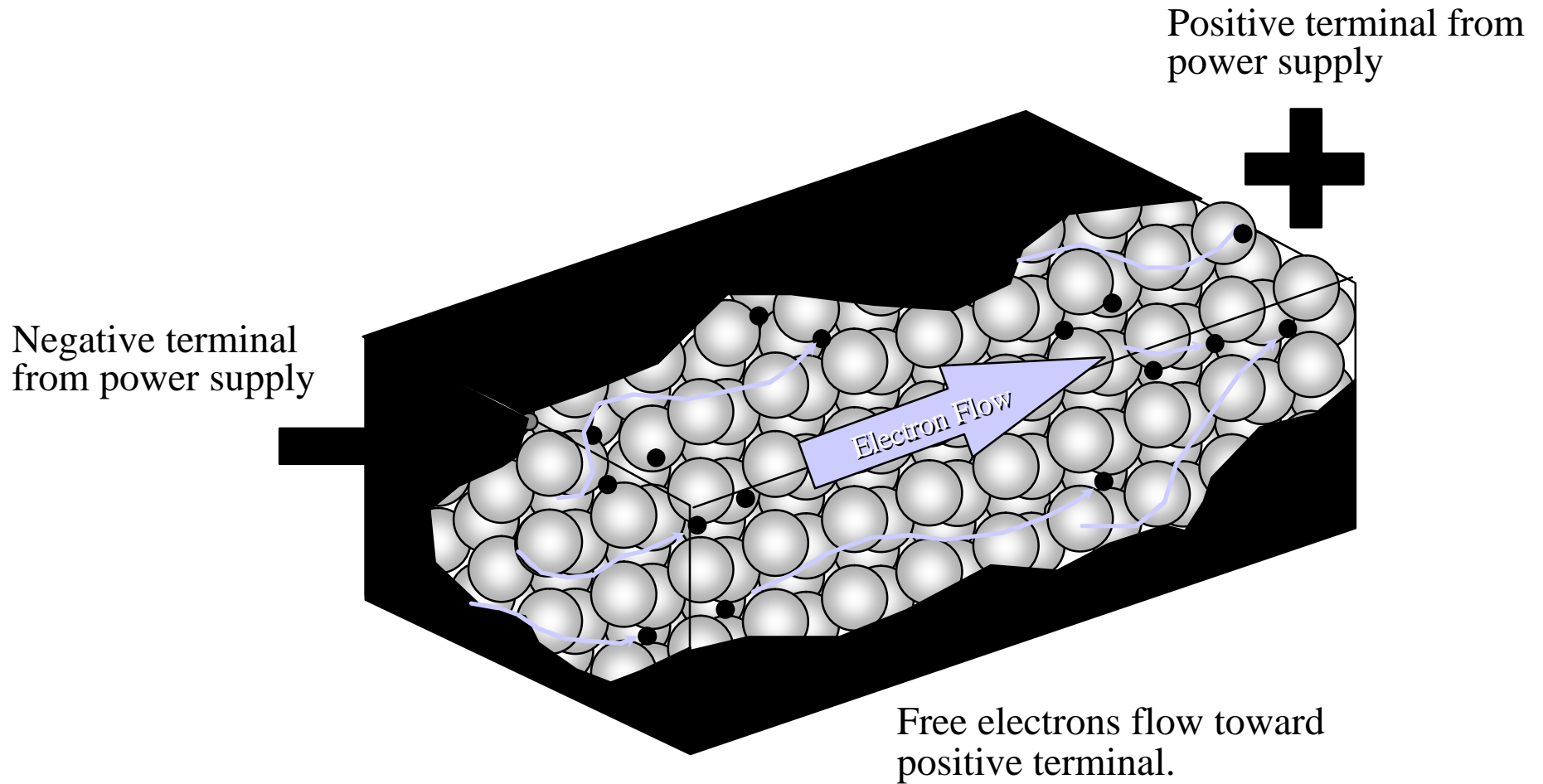
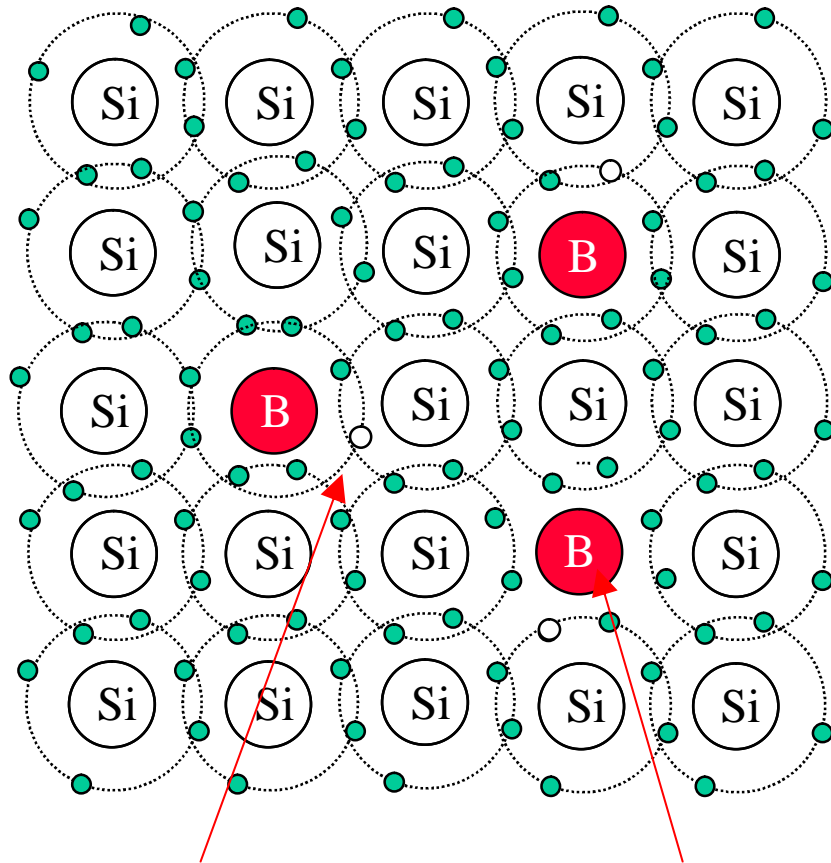


Figure 2.24

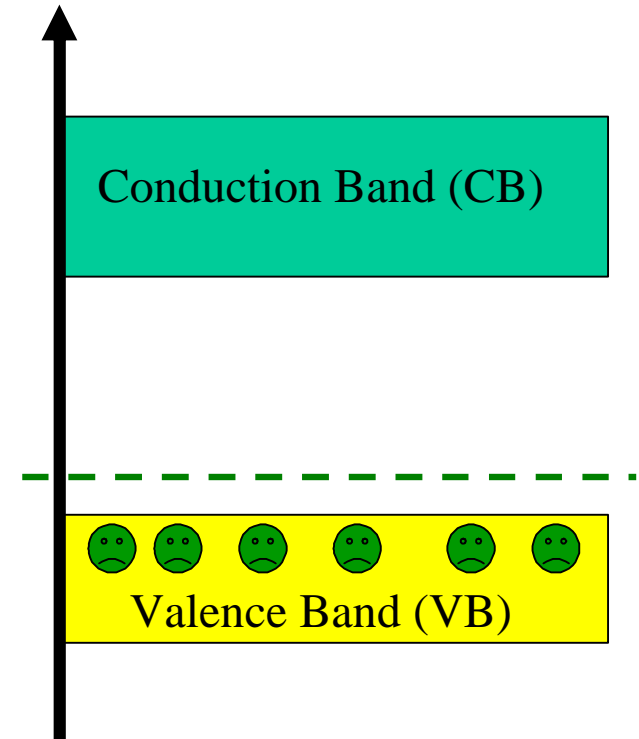


Acceptor atoms provide a deficiency of electrons to form p-type silicon.

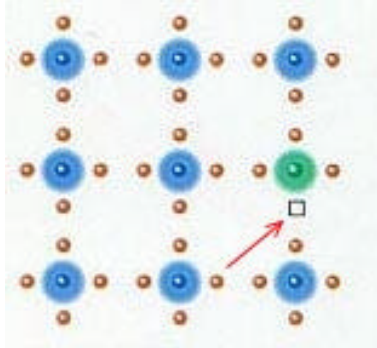


+ Hole

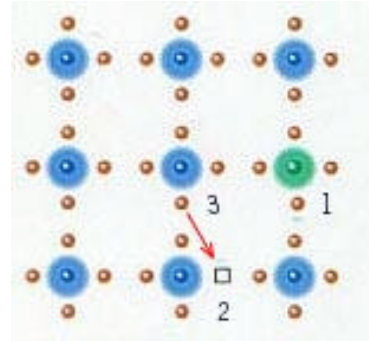
Boron atom serves as p-type dopant



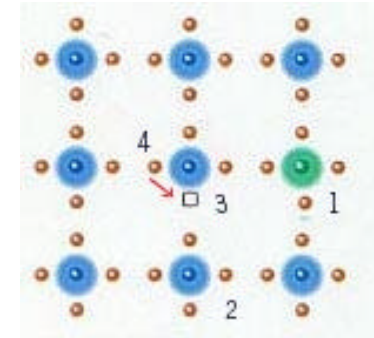
“Hole Movement in Silicon”



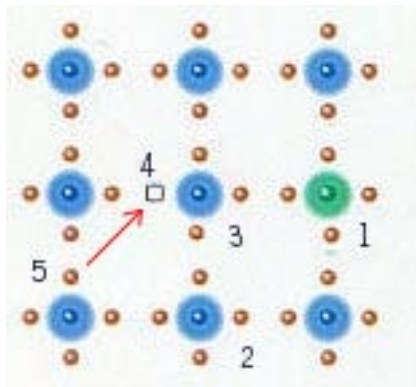
Boron is neutral, but nearby electron may jump to fill bond site.



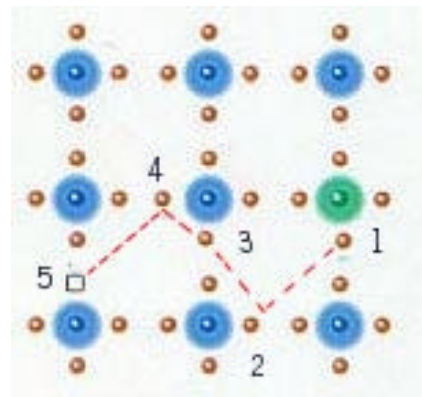
Boron is now a negative ion.



Only thermal energy to kick electrons from atom to atom.



Hole moved from 2 to 3 to 4, and will move to 5.



The empty silicon bond sites (holes) are thought of as being *positive*, since their presence makes that region positive.

Conduction in p-Type Silicon

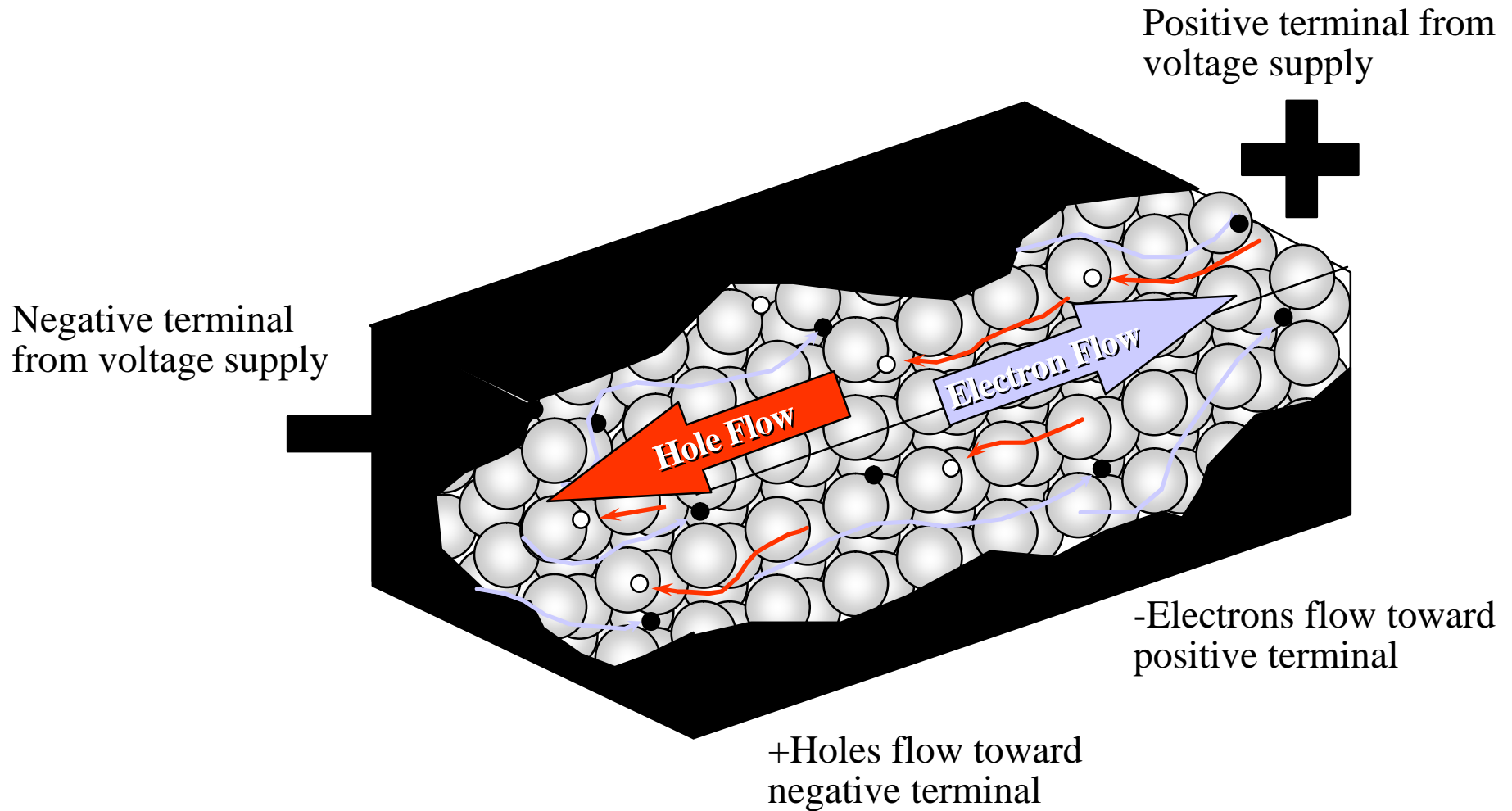


Figure 2.26

Outline

- Silicon and Silicon crystal structure
- Defect types in silicon crystals
- Silicon doping (Dopants = Defects)
- **Radiation induced defects**
 - Point defects and clusters
 - Particle dependence
 - Defect kinetics
 - Example of DLTS measurement
- What are defects doing to detectors?
- How to measure defects?
- Coffee

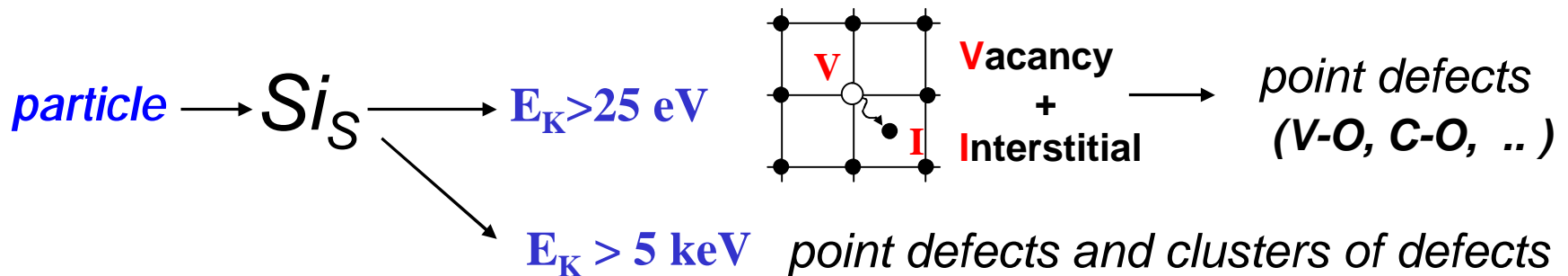
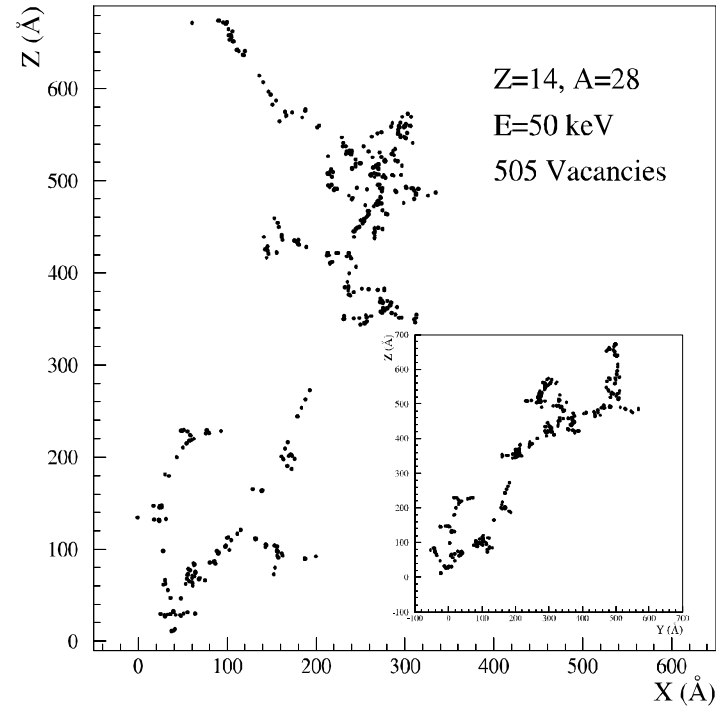
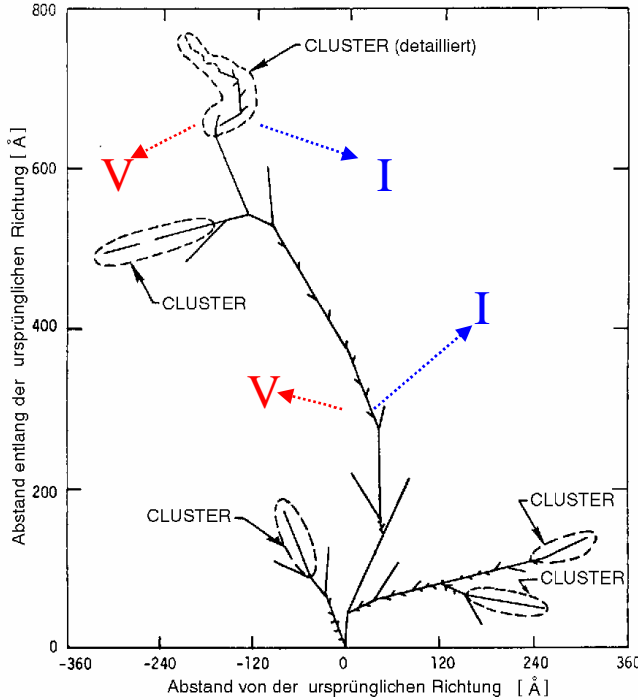


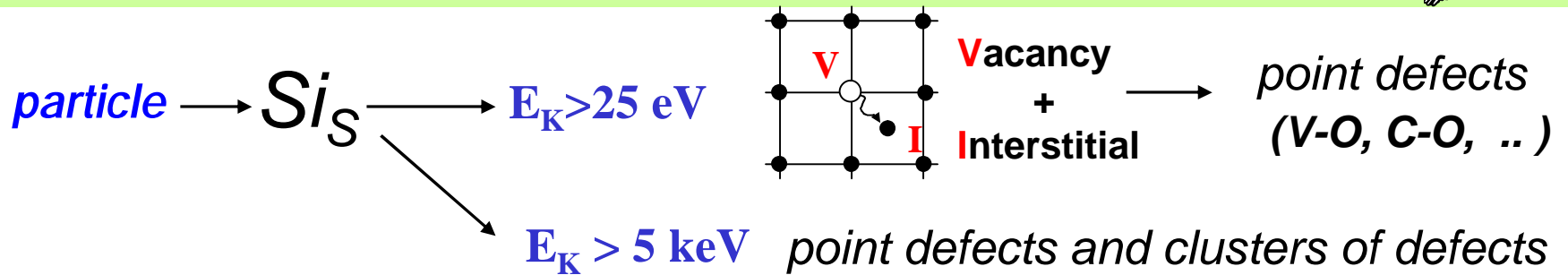


- ◆ Spatial distribution of vacancies created by a 50 keV Si-ion in silicon.
(typical recoil energy for 1 MeV neutrons)

M.Huhtinen 2001

van Lint 1980





• **^{60}Co -gammas**

- Compton Electrons with max. $E_\gamma \approx 1 \text{ MeV}$ (no cluster production)

• **Electrons**

- $E_e > 255 \text{ keV}$ for displacement
- $E_e > 8 \text{ MeV}$ for cluster

• **Neutrons (elastic scattering)**

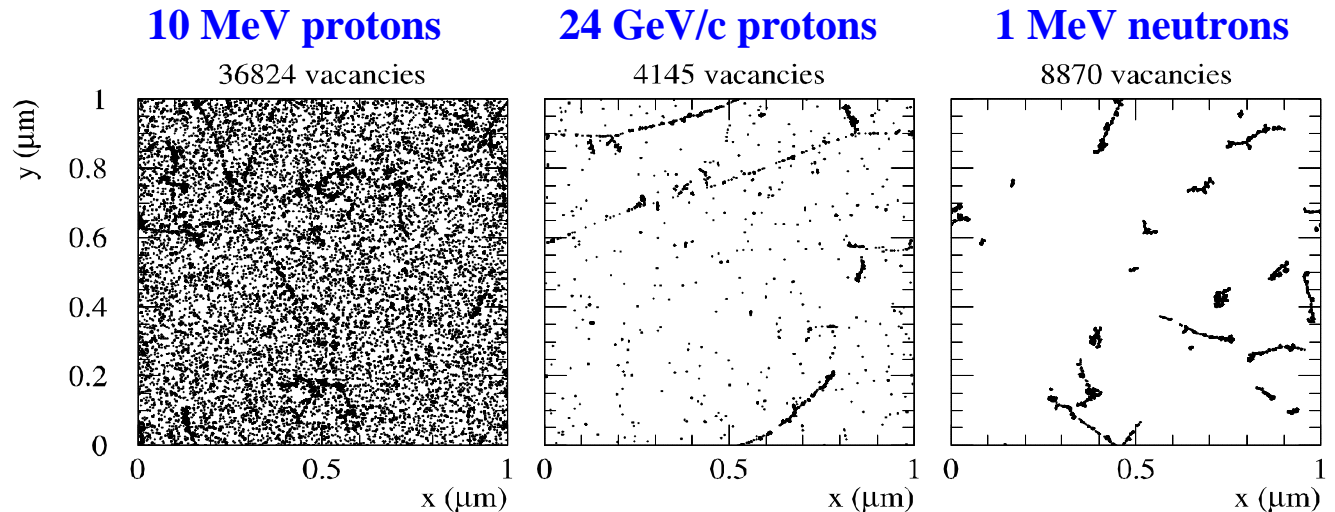
- $E_n > 185 \text{ eV}$ for displacement
- $E_n > 35 \text{ keV}$ for cluster

Only point defects \longleftrightarrow **point defects & clusters** \longleftrightarrow **Mainly clusters**

Simulation:

Initial distribution of vacancies in $(1\text{mm})^3$ after $10^{14} \text{ particles/cm}^2$

[Mika Huhtinen NIMA 491(2002) 194]





- Two basic defects

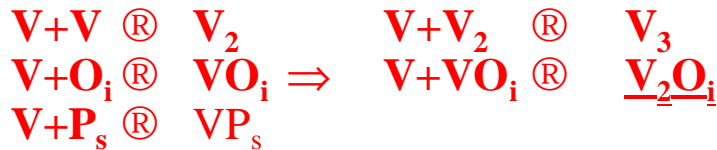
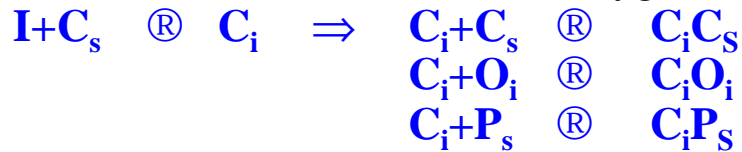
I - Silicon Interstitial V - Vacancy

- Primary defect generation

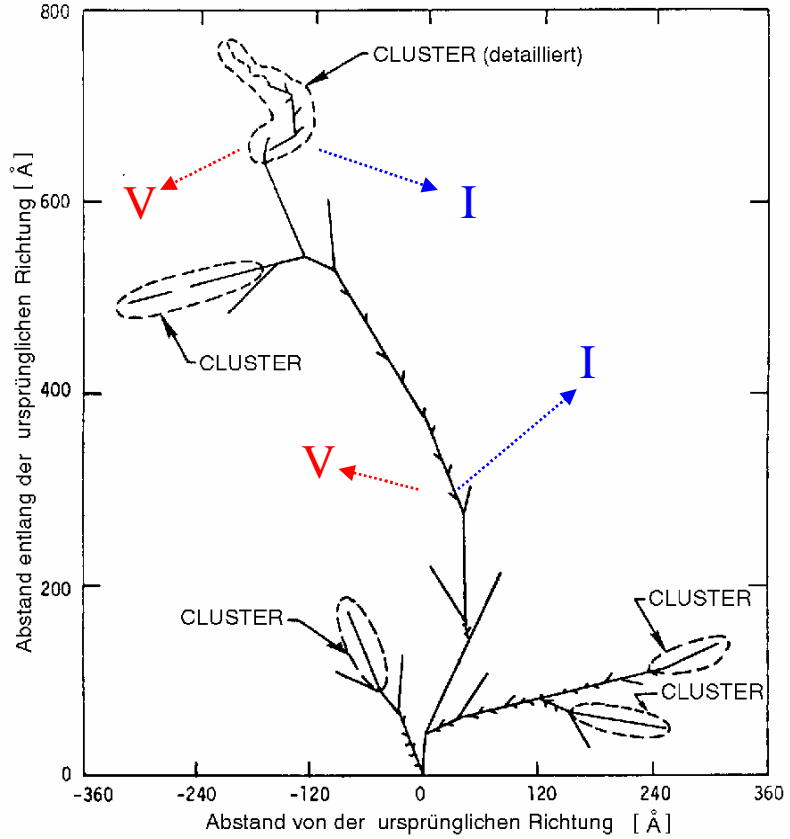
I, I₂ higher order I (?)
 P I-CLUSTER (?) ←
 V, V₂, higher order V (?) **Damage?!**
 P V-CLUSTER (?) ←

- Secondary defect generation

Main impurities in silicon: Carbon (C_s)
 Oxygen (O_i)

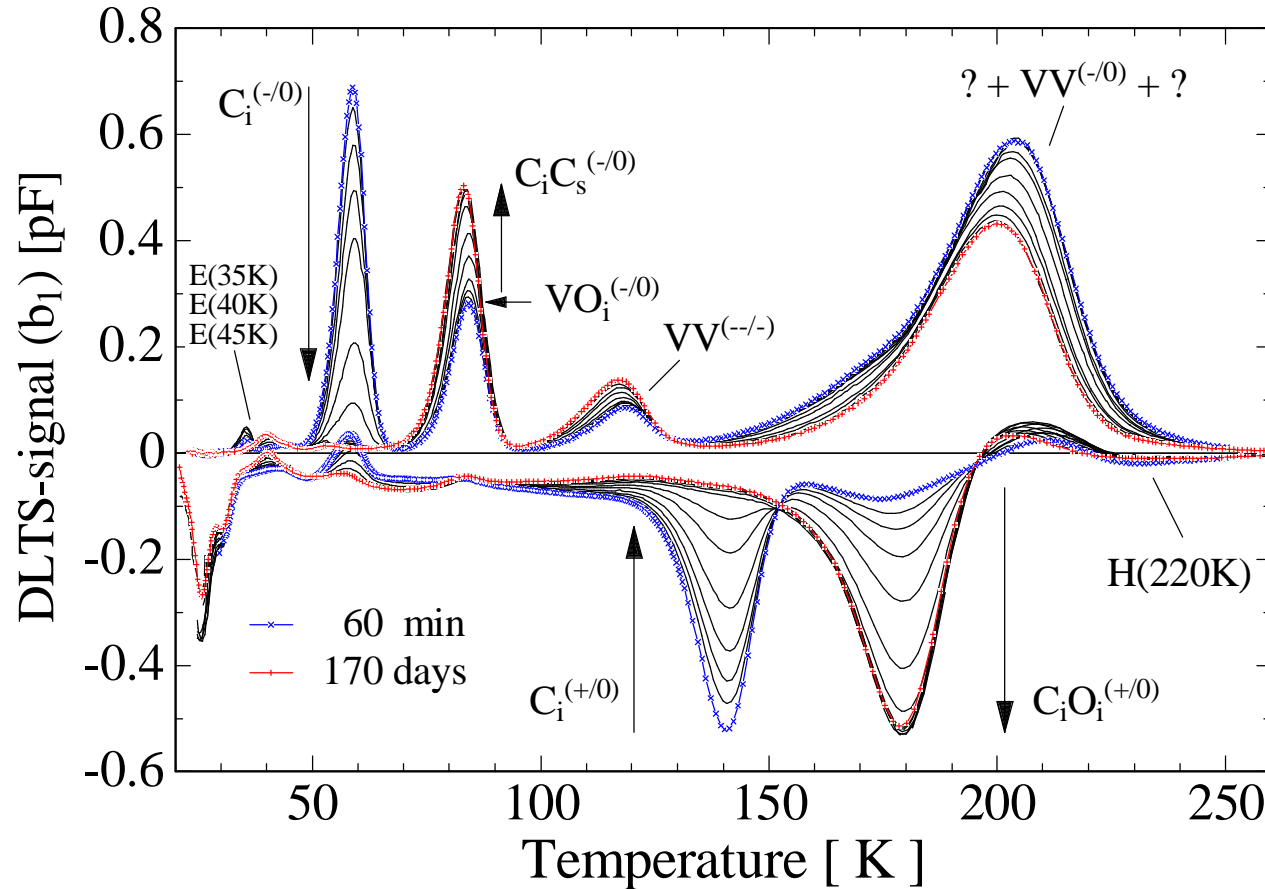


← **Damage?! (“V₂O-model”)**



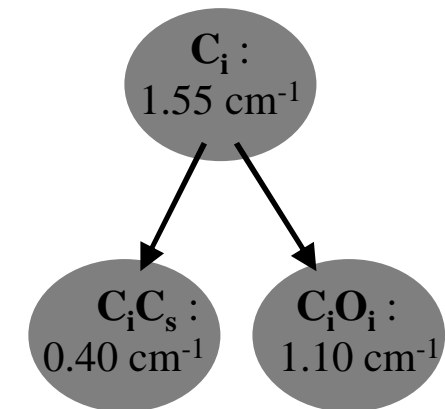


Deep Level Transient Spectroscopy



Introduction Rates

$$N_t / F_{eq}$$



- Introduction rates of main defects $\gg 1 \text{ cm}^{-1}$
- Introduction rate of negative space charge $\gg 0.05 \text{ cm}^{-1}$

example : $F_{eq} = 1 \cdot 10^{14} \text{ cm}^{-2}$

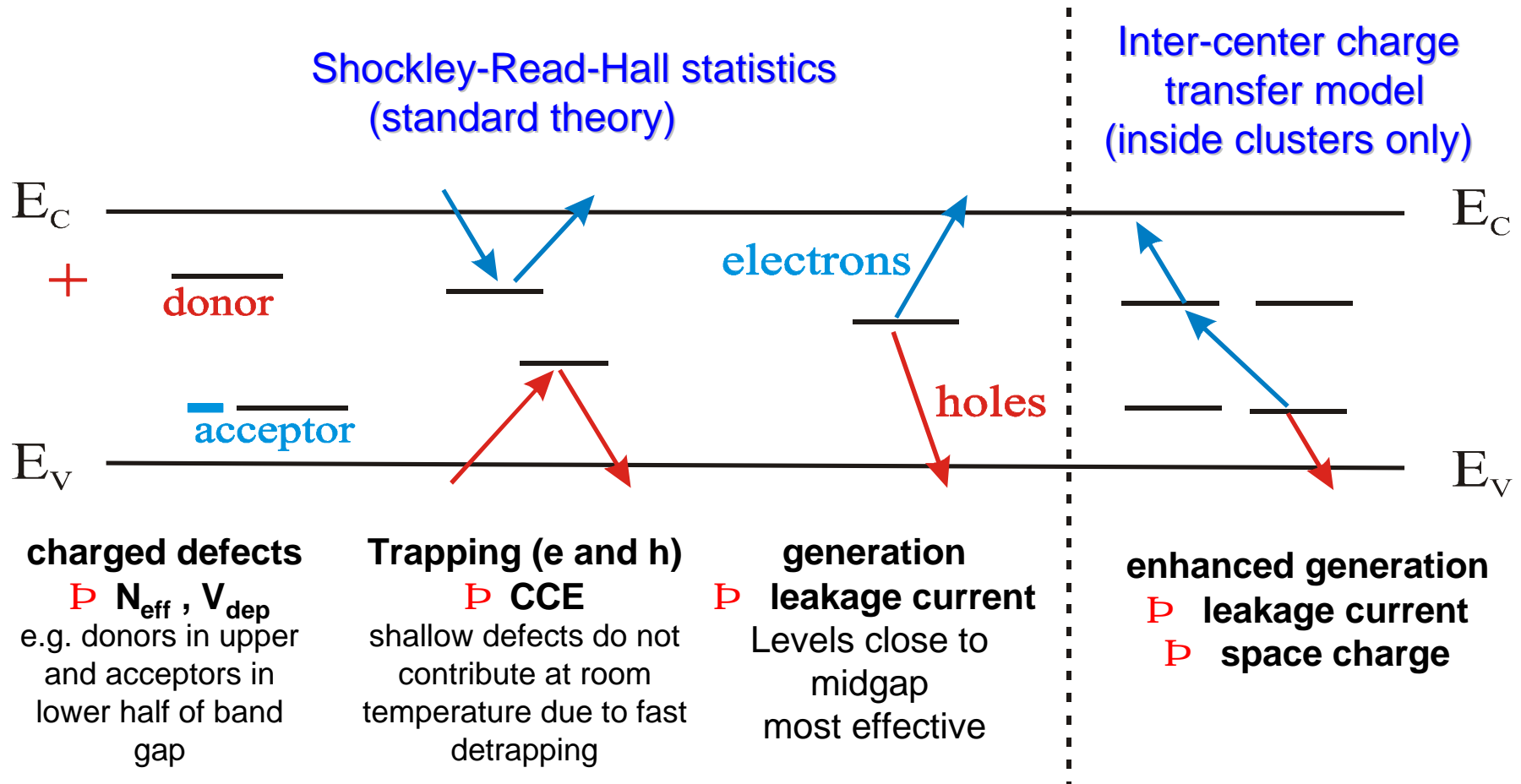
defects $\gg 1 \cdot 10^{14} \text{ cm}^{-3}$

space charge $\gg 5 \cdot 10^{12} \text{ cm}^{-3}$

Outline

- Silicon and Silicon crystal structure
- Defect types in silicon crystals
- Silicon doping (Dopants = Defects)
- Radiation induced defects
- **What are defects doing to detectors?**
- How to measure defects?
- Coffee





Impact on detector properties can be calculated if all defect parameters are known:

$S_{n,p}$: cross sections

DE : ionization energy

N_t : concentration

Outline

- Silicon and Silicon crystal structure
- Defect types in silicon crystals
- Silicon doping (Dopants = Defects)
- Radiation induced defects
- What are defects doing to detectors?
- **How to measure defects?**
 - **Some measurement techniques ...**
 - **Deep Level Transient Spectroscopy**
- Coffee





- **Structure and Chemical Configuration**
 - TEM – Transmission Electron Spectroscopy
 - EPR – Electron Paramagnetic Resonance
- **Optical properties (local vibrational modes)**
 - FTIR – Fourier Transform Infrared spectroscopy
- **Electrical Properties**
 - PL - Photoluminescence
 - TSC – Thermally Stimulated Current
 - **DLTS – Deep Level Transient Spectroscopy**
- **Binding energy and migration**
 - Annealing experiments

CERN Deep Level Transient Spectroscopy

TA1/SSD - simplified working principle -

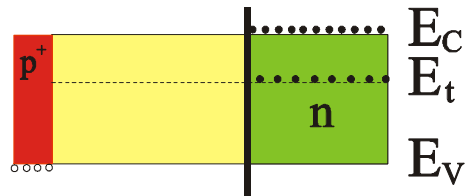


a) Stabilize temperature

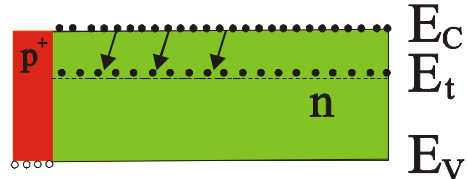
b) Variation of voltage

→ Measurement of capacitance transient

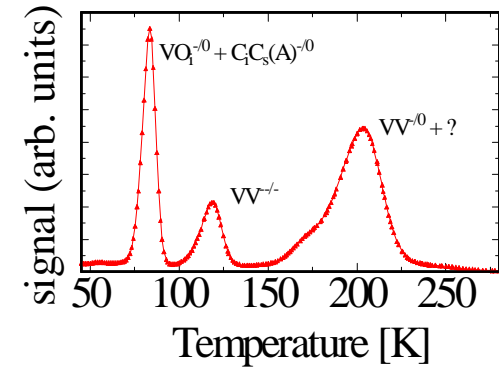
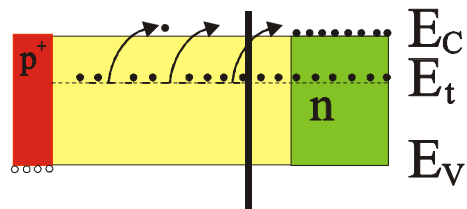
1) reverse bias



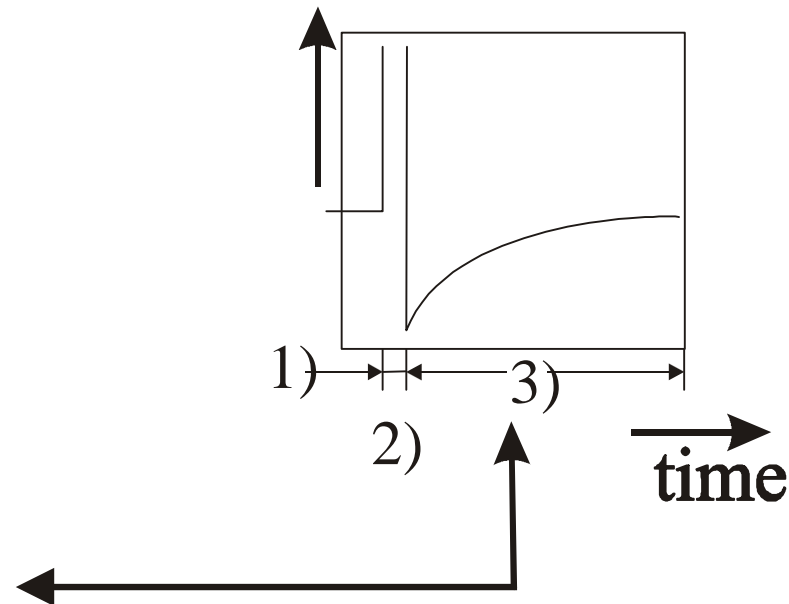
2) zero bias



3) reverse bias



Capacitance

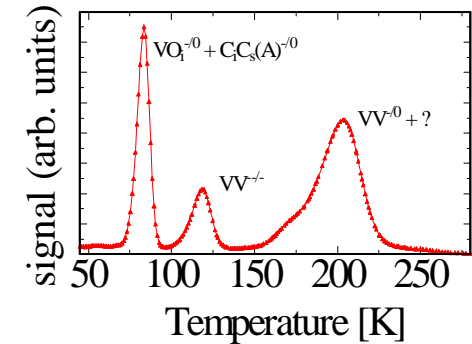
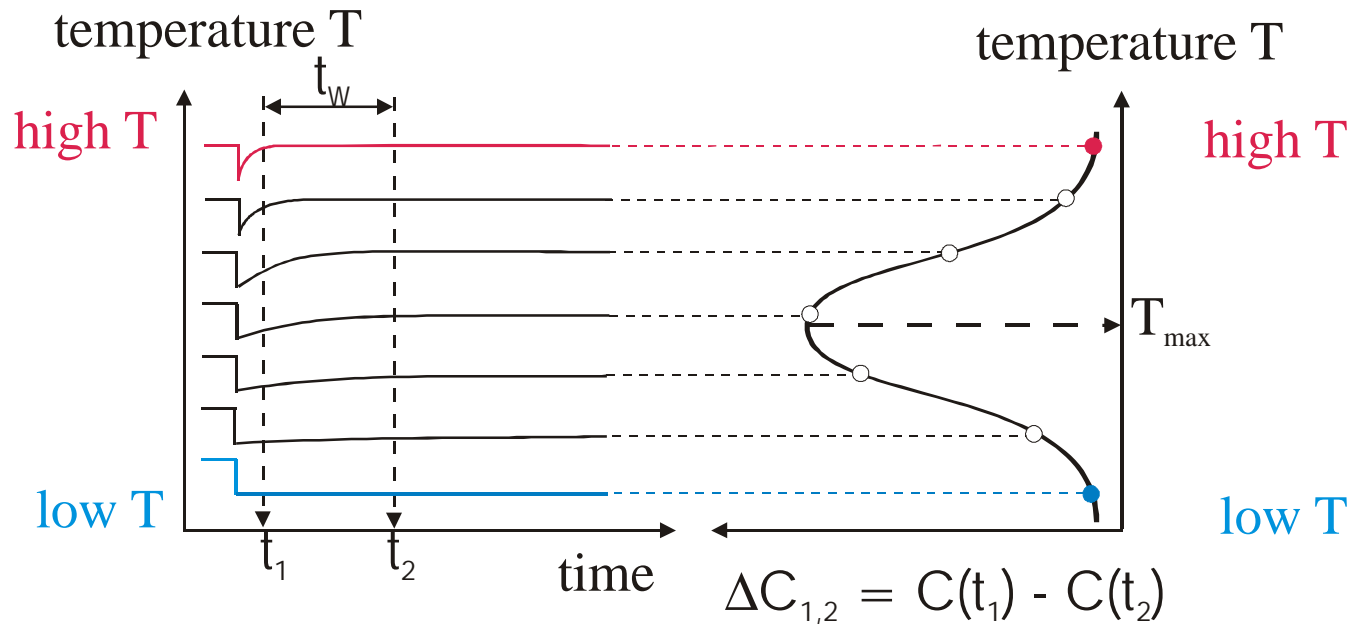


$$C \propto \frac{1}{W} \propto \sqrt{N_{eff}}$$



- simplified working principle -

c) Measure capacitance transients at many temperatures
→ DLTS - spectrum



c) Analyze the DLTS-spectra (“Arrhenius Plots”)

↳ extract defect parameters :

E_t position in bandgap
 S_n, S_p cross sections for electrons and holes
 N_t defect concentration



Coffee !!!!!



Coffee !!!!!

